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Key indicators

Single-crystal X-ray study T = 88 K Mean σ (C–C) = 0.004 Å Disorder in main residue R factor = 0.033 wR factor = 0.065 Data-to-parameter ratio = 9.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2006 International Union of Crystallography All rights reserved In the crystal structure of the title compound, $C_{10}H_{21}NO_5$, molecules are linked by strong $N-H\cdots O$ and $O-H\cdots N$ hydrogen bonds into infinite zigzag chains.

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Comment

This study is part of a programme aimed at generating new hydrogenation catalyst ligands (Gainsford *et al.*, 2006, and references therein). The asymmetric unit of the title compound, (I), contains one independent 1L-1-amino-1-de-oxy-2,3,4,5-tetra-O-methyl-*scyllo*-inositol molecule (Fig. 1). The absolute configuration shown (1R,2S,3R,4R,5S,6R) is that determined from the chemical synthesis.



Two related *scyllo*-inositol compounds have been reported, *viz.* BEKNAG (Solomons *et al.*, 1998) and ZOSSEF (Anderson *et al.*, 1995) [CSD refcodes; Version 5.27, with August 2006 updates; Allen, 2002]. Three others based on *scyllo*-inositol involve fused ring structures, *e.g.* the isopropylidene adduct TERCIC (Lampe *et al.*, 1996). The inositol ring adopts a slightly flattened chair conformation with Q = 0.556 (3) Å, $\theta = 174.4$ (3)° and $\varphi = 168$ (2)° (Cremer & Pople, 1975). The flattening is seen in the ring torsion angles involving C4 [average absolute value is 50.5 (3)°] and was also noted in BEKNAG. One other minor perturbation involves the normally eclipsed methyl on C3, which has a significant twist about the C3–O3 bond; see Table 1 (>18°; Anderson *et al.*, 1995). Other distances and angles are normal.

The crystal structure (Fig. 2) is dominated by strong N– H···O and O–H···N hydrogen bonds, linking molecules by way of an $R_2^2(10)$ motif (Bernstein *et al.*, 1995). A weaker C– H···O interaction provides some linking along the *a* axis (Table 2).

Experimental

1D-3,4,5,6-Tetra-O-methyl-myo-inositol (2.60 g, 11.0 mmol), prepared in a similar manner to the tetra-O-benzyl analogue (Wewers *et al.*, 2005), was dissolved in toluene (130 ml) and dibutyltin(IV) oxide (3.29 g, 13.2 mmol) was added. The solution was heated to reflux using a Dean–Stark apparatus for 2 h and then the solvent was



Figure 1

The molecular structure of (I), with displacement ellipsoids drawn at the 50% probability level.





Packing diagram of (I), viewed down the a axis. Only H atoms involved in hydrogen bonds (dashed lines) are shown. For symmetry codes see Table 2.

removed. Dimethylformamide (80 ml) was added and the solution was cooled to 273 K. Caesium fluoride (3.35 g, 22.0 mmol) and benzyl bromide (2.63 ml, 22.0 mmol) were added; the suspension was stirred at 273 K for 10 min and then at room temperature for 3 h to give 1-*O*-benzyl-3,4,5,6-tetra-*O*-methyl-*myo*-inositol (3.47 g, 97%). The benzyl inositol (3.34 g, 10.2 mmol) was treated with methanesulfonyl chloride (3.99 ml, 51.2 mmol) in pyridine (50 ml) overnight at room temperature to give the mesylate (3.73 g, 90%). The mesylate (3.73 g, 9.23 mmol) was then subjected to a displacement reaction using sodium azide (4.20 g, 64.6 mmol) in dimethylformamide (37 ml) overnight at 383 K to give the azide (2.44 g, 75%). The azide (2.44 g, 7.51 mmol) was stirred in methanol (100 ml) with 10% palladium on carbon (1.00 g) under one atmosphere of hydrogen at room

temperature for 4 d to give the amino alcohol (1.31 g, 80%). Crystals of the title compound were isolated from dichloromethane.

Crystal data

Data collection

Bruker–Nonius APEX2 CCD areadetector diffractometer φ and ω scans Absorption correction: multi-scan (Blessing, 1995) $T_{\rm min} = 0.888, T_{\rm max} = 1.0$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.033$ $wR(F^2) = 0.065$ S = 0.831471 reflections 161 parameters 10832 measured reflections 1471 independent reflections 913 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.064$ $\theta_{\text{max}} = 26.4^{\circ}$

Table 1

Selected geometric parameters (Å, °).

O1-C6	1.431 (3)	N1-C1	1.475 (3)
O1-H10	0.83 (3)	N1-H1N	0.92 (3)
C6-O1-H10	113 (2)	C1-N1-H1N	111.0 (16)
N1-C1-C2-O2	-61.4(3)	C2-C3-C4-C5	49.8 (3)
C6-C1-C2-C3	58.1 (3)	C3-C4-C5-C6	-51.1 (3)
C1-C2-C3-C4	-53.1 (3)	C8-O3-C3-H3	36

Table 2				
Hydrogen-bond	geometry	(Å,	°).	

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N1-H1N\cdots O4^{i}$	0.92 (3)	2.35 (3)	3.258 (3)	171 (2)
O1−H1O···N1 ⁱⁱ	0.83 (3)	2.02 (3)	2.819 (3)	162 (3)
$C10-H10A\cdots O3^{iii}$	0.98	2.59	3.496 (3)	154
	4 -	1	3	

Symmetry codes: (i) $-x + \frac{1}{2}, -y + 2, z - \frac{1}{2}$; (ii) $x + \frac{1}{2}, -y + \frac{3}{2}, -z + 1$; (iii) x + 1, y, z.

In the absence of significant anomalous scattering effects, Friedel pairs were merged. All carbon-bound H atoms were constrained to their expected geometries (methyl C–H = 0.98 Å and methine C–H = 1.00 Å). Amino and hydroxyl H atoms were freely refined, yielding N–H = 0.92 (3)–0.94 (3) Å and O–H = 0.83 (3) Å. All methyl H atoms were free to rotate, except those on C7; the latter were disordered over two sites with refined occupancies of 0.79 (3) and 0.21 (3). $U_{iso}(H) = xU_{eq}(parent atom)$, where x = 1.5 for methyl and x = 1.2 for other H atoms.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT* and *SADABS* (Sheldrick, 1996); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick,

1997); molecular graphics: *ORTEP-3* in *WinGX* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Anderson, J. E., Angyal, S. J. & Craig, D. C. (1995). *Carbohy. Res.* 272, 141–148.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1573.

- Blessing, R. H. (1995). Acta Cryst. A51, 33-38.
- Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Gainsford, G. J., Lensink, C. & Falshaw, A. (2006). Acta Cryst. C62, o650–o652.
 Lampe, D., Liu, C., Mahon, M. F. & Potter, B. V. L. (1996). J. Chem. Soc. Perkin Trans. 1, pp. 1717–1727.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Siemens (1996). *SMART* and *SAINT*. Versions 4.0. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Solomons, K. R. H., Freeman, S., Schwalbe, C. H., Shears, S. B., Nelson, D. J., Xie, W., Bruzik, K. S. & Kaetzel, M. A. (1998). *Carbohydr. Res.* **309**, 337– 343.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Wewers, W., Gillandt, H. & Traub, H. S. (2005). *Tetrahedron Asymmetry*, 16, 1723–1728.